

Section - A (Physics)

1. A tightly wound 100 turns coil of radius 10cm carries a current of 7A. The magnitude of the magnetic field at the centre of the coil is (Take permeability of free space as  $4\pi \times 10^{-7} SI$  units):

- (1) 4.4mT                                      (2) 44T                                      (3) 44mT                                      (4) 4.4T

**Answer: 1**

2. Match List-I with List-II. Choose the correct answer from the options given below:

	Column 1		Column 2
1	Diamagnetic	A	$\chi=0$
2	Ferromagnetic	B	$0 > \chi \geq -1$
3	Paramagnetic	C	$\chi \gg 1$
4	Non- Magnetic	D	$0 < \chi < \epsilon$ (a small positive number)

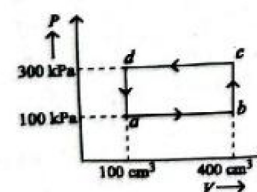
- (1) A-III, B-II, C-I, D-IV                                      (2) A-IV, B-III, C-II, D-I  
 (3) A-II, B-III, C-IV, D-I                                      (4) A-II, B-I, C-III, D-IV

**Answer: 3**

3. A thermodynamic system is taken through the cycle *abcd*. The work done by the gas along the path *bc* is :

- (1) -90 J3                                      (2) -60J                                      (3) zero                                      (4) 30J

**Answer: 3**



4. An unpolarised light beam strikes a glass surface at Brewster's angle. Then

- (1) (1) both the reflected and refracted light will be completely polarised.  
 (2) (2) the reflected light will be completely polarised but the refracted light will be partially polarised.  
 (3) (3) the reflected light will be partially polarised.                                      (4) (4) the refracted light will be completely polarised.

**Answer: 2**

5. In an ideal transformer, the turns ratio is  $\frac{N_p}{N_s} = \frac{1}{2}$ . The ratio  $V_s : V_p$  is equal to (the symbols carry their usual meaning) :

- (1) 1 : 1                                      (2) 1 : 4                                      (3) 1 : 2                                      (4) 2 : 1

**Answer: 4**

6. A logic circuit provides the output  $Y$  as per the following truth table : The expression for the output  $Y$  is :

A	B	Y
0	0	1
0	1	0
1	0	1
1	1	0

- (1)  $\bar{B}$                       (2)  $B$                       (3)  $A \cdot B + \bar{A}$                       (4)  $A \cdot \bar{B} + \bar{A}$

**Answer: 1**

7. In a vernier calipers,  $(N + 1)$  divisions of vernier scale coincide with  $N$  divisions of main scale. If 1 MSD represents  $0.1\text{mm}$ , the vernier constant (in  $\text{cm}$ ) is :

- (1)  $100N$                       (2)  $10(N + 1)$                       (3)  $\frac{1}{10N}$                       (4)  $\frac{1}{100(N+1)}$

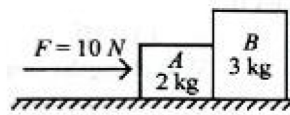
**Answer: 4**

8. The maximum elongation of a steel wire of  $1\text{m}$  length if the elastic limit of steel and its Young's modulus, respectively, are  $8 \times 10^8 \text{Nm}^{-2}$  and  $2 \times 10^{11} \text{Nm}^{-2}$ , is :

- (1)  $40\text{mm}$                       (2)  $8\text{mm}$                       (3)  $4\text{mm}$                       (4)  $0.4\text{mm}$

**Answer: 3**

9. A horizontal force  $10\text{N}$  is applied to a block  $A$  as shown in figure. The mass of blocks  $A$  and  $B$  are  $2\text{kg}$  and  $3\text{kg}$ , respectively. The blocks slide over a frictionless surface. The force exerted by block  $A$  on block  $B$  is :



- (1)  $6\text{N}$                       (2)  $10\text{N}$                       (3) zero                      (4)  $4\text{N}$

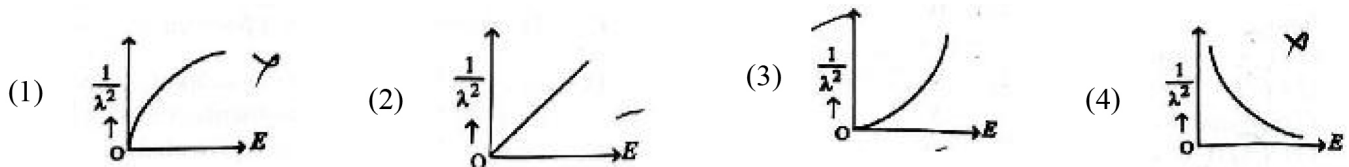
**Answer: 1**

10. If the monochromatic source in Young's double slit experiment is replaced by white light, then

- (1) there will be a central bright white fringe surrounded by a few coloured fringes.  
 (2) all bright fringes will be of equal width.                      (3) interference pattern will disappear.  
 (4) there will be a central dark fringe surrounded by a few coloured fringes.

**Answer: 1**

11. The graph which shows the variation of  $\left(\frac{1}{\lambda^2}\right)^{r-1}$  and its kinetic energy,  $E$  if (where  $\lambda$  is de Broglie wavelength of a free particle):

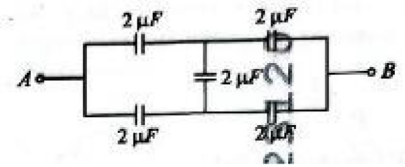


**Answer: 2**

12. In the following circuit, the equivalent capacitance between terminal  $A$  and terminal  $B$  is :

- (1)  $0.5\mu F$       (2)  $4\mu F$       (3)  $2\mu F$       (4)  $1\mu F$

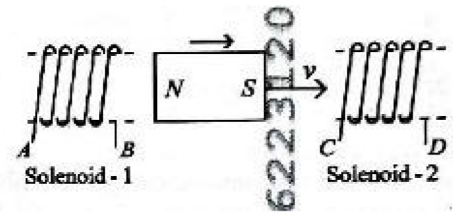
**Answer: 3**



13. In the above diagram, a strong bar magnet is moving towards solenoid-2 from solenoid-1. The direction of induced current in solenoid-1 and that in solenoid-2, respectively, are through the directions:

- (1)  $AB$  and  $CD$  (2)  $BA$  and  $DC$  (3)  $AB$  and  $DC$  (4)  $BA$  and  $CD$

**Answer: 3**



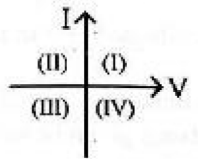
14. Consider the following statements A and B and identify the correct answer :

A. For a solar-cell, the I-V characteristics lies in the IV quadrant of the given graph.

B. In a reverse biased pn junction diode, the current measured ( $\mu A$ ), is due to majority charge carriers.

- (1) Both  $A$  and  $B$  are correct.      (2) Both  $A$  and  $B$  are incorrect, 1)?  
 (3)  $A$  is correct but  $B$  is incorrect.      (4)  $A$  is incorrect but  $B$  is correct.

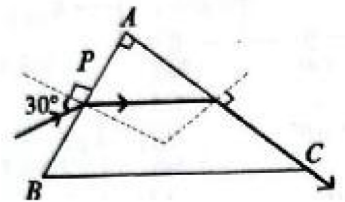
**Answer: 3**



15. A light ray enters through a right angled prism at point  $P$  with the angle of incidence  $30^\circ$  as shown in figure. It travels through the prism parallel to its base  $BC$  and emerges along the face  $AC$ . The refractive index of the prism is:

- (1)  $\frac{\sqrt{3}}{4}$       (2)  $\frac{\sqrt{3}}{2}$       (3)  $\frac{\sqrt{5}}{4}$       (4)  $\frac{\sqrt{5}}{2}$

**Answer: 4**



16. Given below are two statements: one is labelled as Assertion  $A$  and the other is labelled as Reason  $R$ .

Assertion  $A$  : The potential ( $V$ ) at any axial point, at  $2m$  distance ( $r$ ) from the centre of the dipole of dipole moment vector  $\vec{P}$  of magnitude,  $4 \times 10^{-6} Cm$ , is  $\pm 9 \times 10^3 V$ .

(Take  $\frac{1}{4\pi\epsilon_0} = 9 \times 10^9$  SI units)

Reason  $R$  :  $V = \pm \frac{2P}{4\pi\epsilon_0 r^2}$ , where  $r$  is the distance of any axial point, situated at  $2m$  from the centre of the dipole.

In the light of the above statements, choose the correct answer from the options given below:

- (1)  $A$  is true but  $R$  is false.      (2)  $A$  is false but  $R$  is true.  
 (3) Both  $A$  and  $R$  are true and  $R$  is the correct explanation of  $A$ .  
 (4) Both  $A$  and  $R$  are true and  $R$  is NOT the correct explanation of  $A$ .

**Answer: 1**

17. The moment of inertia of a thin rod about an axis passing through its mid point and perpendicular to the rod, is  $2400 gm^2$ . The length of the  $400g$  rod is nearly"

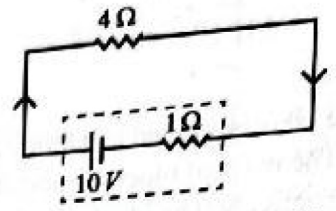
- (1)  $20.7cm$       (2)  $72.0 cm$       (3)  $8.5 cm$       (4)  $17.5 cm$

**Answer: 3**

18. The terminal voltage of the battery, whose emf is  $10V$  and internal resistance  $1\Omega$ , when connected through an external resistance of  $4\Omega$  as shown in the figure is :

- (1)  $8V$                       (2)  $10V$                       (3)  $4V$                       (4)  $6V$

**Answer: 1**



19. Match List I with List II. Choose the correct answer from the options given below:

- (1) A-IV, B-III, C-I, D-II      (2) A-I, B-II, C-III, D-IV      (3) A-II, B-I, C-IV, D-III      (4) A-III, B-IV, C-II, D-I

**Answer: 4**

20. If  $c$  is the velocity of light in free space, the correct statements about photon among the following are :

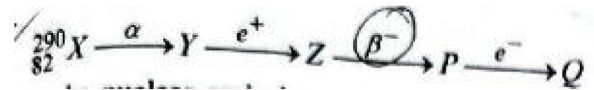
- A. The energy of a photon is  $E = hv$ .  
 B. The velocity of a photon is  $c$ .  
 C. The momentum of a photon,  $p = \frac{hv}{c}$ .  
 D. In a photon-electron collision, both total energy and total momentum are conserved.  
 E. Photon possesses positive charge.

Choose the correct answer from the options given below:

- (1) A C and D only                      (2) A B, D and E only                      (3) A and B only                      (4) A, B, C and D only

**Answer: 4**

21. In the nuclear emission stated above, the mass number and atomic number of the product  $Q$  respectively, are :



- (1) 288,82                      (2) 286,81                      (3) 280,81                      (4) 286,80

**Answer: 2**

22. At any instant of time  $t$ , the displacement of any particle is given by  $2t - 1$  (SI unit) under the influence of force of  $5N$ . The value of instantaneous power is (in SI unit):

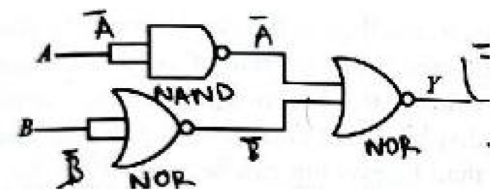
- (1) (1) 7                      (2) (2) 6                      (3) (3) 10                      (4) (4) 5

**Answer: 3**

23. The output ( $Y$ ) of the given logic gate is similar to the output of an/a :

- (1) (1) OR gate                      (2) (2) AND gate  
 (3) (3) NAND gate                      (4) (4) NOR gate

**Answer: 1**



24. The mass of a planet is  $\frac{1}{10}$  th that of the earth and its diameter is half that of the earth. The acceleration due to gravity on that planet is :

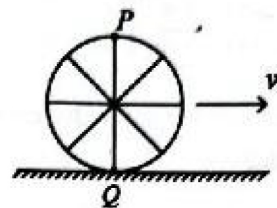
- (1)  $4.9ms^{-2}$                       (2)  $3.92ms^{-2}$                       (3)  $19.6ms^{-2}$                       (4)  $9.8ms^{-2}$

**Answer: 2**

25. Given below are two statements : Statement I : Atoms are electrically neutral as they contain equal number of positive and . negative charges. Statement II : Atoms of each element are stable and emit their characteristic spectrum. In the fight of the above statements, choose the most appropriate answer from the options given below :
- (1) (1) Statement I is correct but Statement II is incorrect.(2) (2) Statement I is incorrect but Statement II is correct.  
(3) (3) Both Statement I and Statement II are correct. (4) (4). Both Statement I and Statement II are incorrect.

**Answer: 1**

26. A wheel of a bullock cart is rolling on a level road as shown in the figure below. If its linear speed is  $v$  in the direction shown, which one of the following options is correct (  $P$  and  $Q$  are any highest and lowest points on the wheel, respectively)?



- (1) Both the points  $P$  and  $Q$  move with equal speed.(2) Point  $P$  has zero speed.  
(3) Point  $P$  moves slower than point  $Q$ . (4) Point  $P$  moves faster than point  $Q$ .

**Answer: 4**

27. A particle moving with uniform speed in a circular path maintains :

- (1) constant velocity but varying acceleration. (2) varying velocity and varying acceleration. (3) constant velocity.  
(4) constant acceleration.

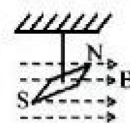
**Answer: 2**

28. A thin flat circular disc of radius  $4.5\text{cm}$  is placed gently over the surface of water. If surface tension of water is  $0.07\text{Nm}^{-1}$ , then the excess force required to take it away from the surface is :

- (1)  $1.98\text{mN}$  (2)  $99\text{N}$  (3)  $19.8\text{mN}$  (4)  $198\text{N}$

**Answer: 3**

29. In a uniform magnetic field of  $0.049\text{T}$ , a magnetic needle performs 20 complete oscillations in 5 seconds as shown. The moment of inertia of the needle is  $9.8 \times 10^{-6}\text{kgm}^2$ . If the magnitude of magnetic moment of the needle is  $x \times 10^{-5}\text{Am}^2$ ; then the value of '  $x$  ' is :



- (1)  $50\pi^2$  (2)  $1280\pi^2$  (3)  $5\pi^2$  (4)  $128\pi^2$

**Answer: 2**

30. Two bodies A and B of same mass undergo completely inelastic one dimensional collision. The body A moves with velocity  $v_1$  while body  $B$  is at rest before collision. The iverelocity of the system after collision is  $v_2$ . The ratio  $v_1 : v_2$  is :

- (1)  $4 : 1$  (2)  $1 : 4$  (3)  $1 : 2$  (4)  $2 : 1$

**Answer: 4**

31. If  $x = 5 \sin\left(\pi t + \frac{\pi}{3}\right)\text{m}$  represents the motion of a particle executing simple harmonic motion, the amplitude and time period of motion, respectively, are :

- (1)  $5\text{cm}, 1\text{s}$  (2)  $5\text{m}, 1\text{s}$  (3)  $5\text{cm}, 2\text{s}$  (4)  $5\text{m}, 2\text{s}$

**Answer: 4**

32. The quantities which have the same dimensions as those of solid angle are :

- (1) strain and arc. (2) angular speed and stress (3) strain and angle (4) stress and angte

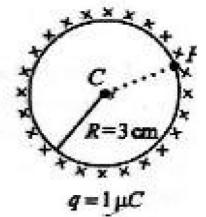
**Answer: 3**

33. A thin spherical shell is charged by some source. The potential difference between the two points  $C$  and  $P$  (in  $V$ ) shown in the figure is:

(Take  $\frac{1}{4\pi\epsilon_0} = 9 \times 10^9$  SI units)

- (1)  $0.5 \times 10^5$                       (2) zero                      (3)  $3 \times 10^5$                       (4)  $1 \times 10^5$

**Answer: 2**



34. A bob is whirled in a horizontal plane by means of a string with an initial speed of  $\omega$  rpm. The tension in the string is  $T$ . If speed becomes  $2\omega$  while keeping the same radius, the tension in the string becomes :

- (1)  $\frac{T}{4}$                       (2)  $\sqrt{2}T$                       (3)  $T$                       (4)  $4T$

**Answer: 4**

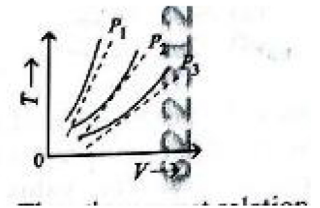
35. A wire of length '  $r$  ' and resistance  $100\Omega$  is divided into 10 equal parts. The first 5 parts are connected in series while the next 5 parts are connected in parallel. The two combinations are again connected in series. The resistance of this final combination is:

- (1)  $55\Omega$  -                      (2)  $60\Omega$   $\propto$                       (3)  $26\Omega$   $\propto$                       (4)  $52\Omega$  -

**Answer: 4**

### Section - B (Physics)

36. The following graph represents the  $T$ - $V$  curves of an ideal gas (where  $T$  is the temperature and  $V$  the volume) at three pressures  $P_1, P_2$  and  $P_3$  compared with those of Charles's law represented as dotted lines,



Then the correct relation is:

- (1)  $P_2 > P_1 > P_3$  > (2)  $P_1 > P_2 > P_3$  (3)  $P_3 > P_2 > P_1$  (4)  $P_1 > P_3 > P_2$  >

**Answer: 2**

37. A parallel plate capacitor is charged by connecting it to a battery through a resistor. If  $I$  is the current in the circuit then in the gap between the plates:

- (1) displacement current of magnitude equal to  $I$  flows in a direction opposite to that of  $I$ .  
 (2) displacement current of magnitude greater than  $I$  flows but can be in any direction.      (3) there is no current.  
 (4) displacement current of magnitude equal to  $I$  flows in the same direction as  $I$ .

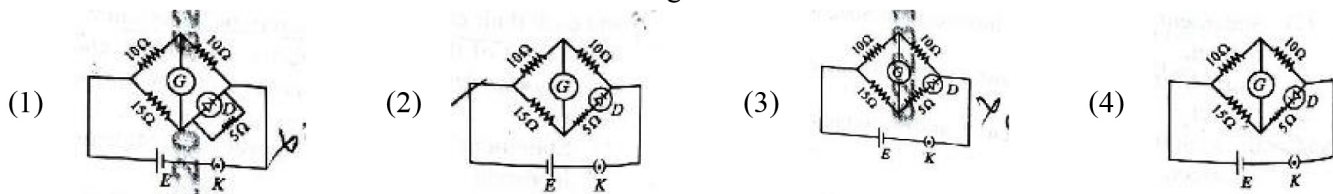
**Answer: 4**

38. The property which is not of an electromagnetic wave travelling in free space is that :

- (1) they travel with a speed equal to  $\frac{1}{\sqrt{\mu_0\epsilon_0}}$ .      (2) they originate from charges moving with uniform speed.  
 (3) they are transverse in nature. (4) the energy density in electric field is equal to energy density in magnetic field.

**Answer: 2**

39. Choose the correct circuit which can achieve the bridge balance.



**Answer: 3**

40. If the plates of a parallel plate capacitor connected to a battery are moved close to each other, then
- A. the charge stored in it, increases. B. the energy stored in it, decreases. C. its capacitance increases. D. the ratio of charge to its potential remains the same. E. the product of charge and voltage increases. Choose the most appropriate answer from the options given below:

(1) B, D and E only      (2) A, B and C only      (3) A, B and E only      (4) A, C and E only

**Answer: 4**

41. A force defined by  $F = \alpha t^2 + \beta t$  acts on a particle at a given time  $t$ . The factor which is dimensionless, if  $\alpha$  and  $\beta$  are constants, is:

(1)  $\alpha\beta t$       (2)  $\alpha\beta/t$       (3)  $\beta t/\alpha$       (4)  $\alpha t/\beta$

**Answer: 4**

42. A metallic bar of Young's modulus,  $0.5 \times 10^{11} \text{ Nm}^{-2}$  and coefficient of linear thermal expansion  $10^{-5} \text{ }^\circ\text{C}^{-1}$  length  $1 \text{ m}$  and area of cross-section  $10^{-3} \text{ m}^2$  is heated from  $0^\circ\text{C}$  to  $100^\circ\text{C}$  without expansion or bending. The compressive force developed in it is:

(1)  $100 \times 10^3 \text{ N}$       (2)  $2 \times 10^3 \text{ N}$       (3)  $5 \times 10^3 \text{ N}$       (4)  $50 \times 10^3 \text{ N}$

**Answer: 4**

43. A small telescope has an objective of focal length  $140 \text{ cm}$  and an eye piece of focal length  $5.0 \text{ cm}$ . The magnifying power of telescope for viewing a distant object is:

(1) 17      (2) 32      (3) 34      (4) 28

**Answer: 4**

44. An iron bar of length  $L$  has magnetic moment  $M$ . It is bent at the middle of its length such that the two arms make an angle  $60^\circ$  with each other. The magnetic moment of this new magnet is:

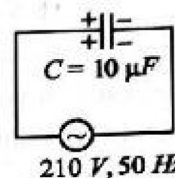
(1)  $2M$       (2)  $\frac{M}{\sqrt{3}}$       (3)  $M$       (4)  $\frac{M}{2}$

**Answer: 4**

45. A  $10 \mu\text{F}$  capacitor is connected to a  $210 \text{ V}$ ,  $50 \text{ Hz}$  source as shown in figure. The peak current in the circuit is nearly ( $\pi = 3.14$ ):

(1)  $1.20 \text{ A}$       (2)  $0.35 \text{ A}$       (3)  $0.58 \text{ A}$       (4)  $0.93 \text{ A}$

**Answer: 4**



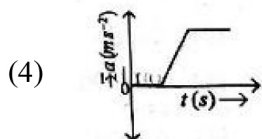
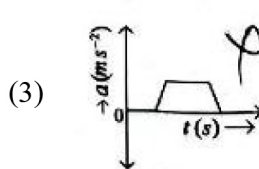
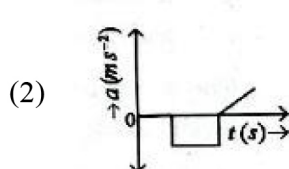
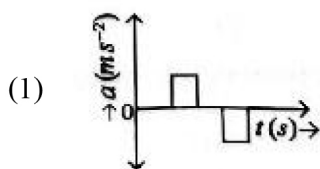
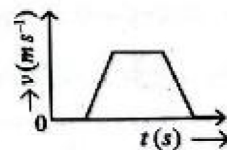
46. Two heaters  $A$  and  $B$  have power rating of  $1 \text{ kW}$  and  $2 \text{ kW}$ , respectively. Those two are first connected in series and then in parallel to a fixed power source. The ratio of power outputs for these two cases is:

(1) 1 : 2      (2) 2 : 3      (3) 1 : 1      (4) 2 : 9

**Answer: 4**

47. The velocity ( $v$ ) - time ( $t$ ) plot of the motion of a body is shown below :

The acceleration ( $a$ ) -time ( $t$ ) graph that best suits this motion is :



**Answer: 1**

48. If the mass of the bob in a simple pendulum is increased to thrice its original mass and its length is made half its original length, then the new time period of oscillation is  $\frac{x}{2}$  times its original time period. Then the value of  $x$  is:

(1)  $2\sqrt{3}$

(2) 4

(3)  $\sqrt{3}y$

(4)  $\sqrt{2}$

**Answer: 4**

49. The minimum energy required to launch a satellite of mass  $m$  from the surface of earth of mass  $M$  and radius  $R$  in a circular orbit at an altitude of  $2R$  from the surface of the earth is:

(1)  $\frac{GmM}{2R}$

(2)  $\frac{GmM}{3R}$

(3)  $\frac{5GmM}{6R}$

(4)  $\frac{2GmM}{3R}$

**Answer: 3**

50. A sheet is placed on a horizontal surface in front of a strong magnet's pole. A force is needed to :

A. hold the sheet there if it is magnetic.

B. hold the sheet there if it is non-magnetic.

C. move the sheet away from the pole with uniform velocity if it is conducting.

D. move the sheet away from the pole with uniform velocity if it is both, non-conducting and non-polar.

Choose the correct statement(s) from the options given below:  $N$

(1) A, C and D only

(2) C only

(3) B and D only

(4) A and C only

**Answer: 1**

**Section - A (Chemistry)**

51. Match the Following Question Choose the correct answer from the options given below:

	List 1		List 2
A	1 mol of $H_2O$ to $O_2$	I	3F
B	1 mol of $MnO_4^-$ to $Mn^{2+}$ in N	II	2F
C	1.5 mol of $9$ from molten $CaCl_2$	III	1F
D	1 mol of $FeO$ to $Fe_2O_3$	IV	5F

(1) A-II, B-III, C-I, D-IV

(2) A-III, B-IV, C-II, D-I

(3) A-II, B-IV, C-I, D-III

(4) A-III, B-IV, C-I, D-II

**Answer: 4**

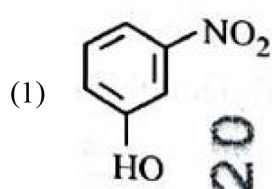


52. Which reaction is NOT a redox reaction?

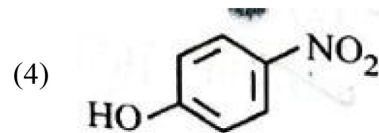
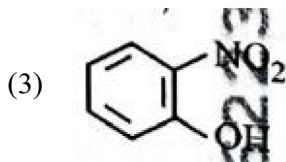
- (1)  $H_2 + Cl_2 \rightarrow 2HCl$  (2)  $BaCl_2 + Na_2SO_4 \rightarrow BaSO_4 + 2NaCl$  (3)  $Zn + CuSO_4 \rightarrow ZnSO_4 + Cu$   
 (4)  $2KClO_3 \rightarrow 2KCl + O_2$

**Answer: 2**

53. Intramolecular hydrogen bonding is present in



(2)  $HF$



**Answer: 3**

54. Fehling's solution 'A' is

- (1) alkaline solution of sodium potassium tartrate (Rochelle's salt) (2) aqueous sodium citrate  
 (3) aqueous copper sulphate (4) alkaline copper sulphate

**Answer: 3**

55. 1 g of sodium hydroxide was treated with 25 mL of 0.75 M HCl solution, the mass of sodium hydroxide left unreacted is equal to

- (1) Zero mg (2) 200 mg (3) 750 mg (4) 250 mg

**Answer: 3**

56. Match The Following

	List 1		List 2
A	$NH_3$	I	Trigonal Pyramidal
B	$BrF_5$	II	Square Planar
C	$XeF_4^-$	III	Octahedral
D	$SF_6$	IV	Square Pyramidal

(1) A-III, B-IV, C-I, D-II

(2) A-II, B-III, C-IV, D-I

(3) A-I, B-IV, C-II, D-III

(4) A-II, B-IV, C-III, D-I

**Answer: 3**

57. The  $E^\circ$  value for the  $Mn^{3+}/Mn^{2+}$  couple is more positive than that of  $Cr^{3+}/Cr^{2+}$  or  $Fe^{3+}/Fe^{2+}$  due to change of

- (1)  $d^4$  to  $d^5$  configuration (2)  $d^3$  to  $d^5$  configuration- (3)  $d^5$  to  $d^4$  configuration (4)  $d^5$  to  $d^2$  configuration

**Answer: 1**

58. Match List I with List II. Choose the correct answer from the options given below:

	List 1		List 2
A	Isothermal process	I	No Heat Exchange
B	Isochoric Process	II	Carried out at constant Temperature
C	Isobaric Process	III	Carried out at constant volume
D	Adiabatic Process	IV	Carried out at constant pressure

(1) A-I, B-II, C-III, D-IV

(2) A-II, B-III, C-IV, D-I

(3) A-IV, B-III, C-II, D-I

(4) A-IV, B-II, C-III, D-I

**Answer: 2**

59. Activation energy of any chemical reaction can be calculated if one knows the value of

(1) orientation of reactant molecules during collision.

(2) rate constant at two different temperatures.

(3) rate constant at standard temperature.

(4) probability of collision.

**Answer: 2**

60. A compound with a molecular formula of  $C_6H_{14}$  has two tertiary carbons. Its IUPAC name is:

(1) 2,3-dimethylbutane

(2) 2,2-dimethylbutane

(3) n-hexane

(4) 2-methylpentane

**Answer: 1**

61. 'Spin only' magnetic moment is same for which of the following ions?

A.  $Ti^{3+}$

C.  $Mn^{2+}$

B.  $Cr^{2+}$

E.  $Sc^{3+}$  Choose the most appropriate answer from the options given below: 6

(1) B and C only

(2) A and D only

(3) B and D only

(4) A and E only

**Answer: 1**

62. Arrange the following elements in increasing order of electronegativity:

$N, O, F, C, Si$  Choose the correct answer from the options given below:

(1)  $O < F < N < C < Si >$

(2)  $F < O < N < C < Si <$

(3)  $Si < C < N < O < F <$

(4)  $Si < C < O < N < F <$

**Answer: 3**

63. Which one of the following alcohols reacts instantaneously with lucas reagent?

(1) 1

(2) 2

(3) 3

(4) 4

**Answer: 2**

64. Given below are two statements :

Statement I : Both  $[Co(NH_3)_6]^{3+}$  and  $[CoF_6]^{3-}$  complexes are octahedral but differ in their magnetic behaviour.

Statement II :  $[Co(NH_3)_6]^{3+}$  is diamagnetic whereas  $[CoF_6]^{3-}$  is paramagnetic. 000, 10, 16. In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I is true but Statement II is false.                      (2) Statement I is false but Statement II is true.  
 (3) Both Statement I and Statement II are true.                      (4) Both Statement I and Statement II are false.

**Answer: 3**

65. Given below are two statements:

Statement I : The boiling point of hydrides of Group 16 elements follow the order



Statement II : On the basis of molecular mass,  $H_2O$  is expected to have lower boiling point than the other members of the group but due to the presence of extensive  $H$ -bonding in  $H_2O$ , it has higher boiling point. In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I is true but Statement II is false.                      (2) Statement I is false but Statement II is true.  
 (3) Both Statement I and Statement II are true.                      (4) Both Statement I and Statement II are false.

**Answer: 3**

66. Match List I with List II. Choose the correct answer from the options given below:

	List 1		List 2
A	$m_l$	I	Shape of orbital
B	$m_s$	II	Size of Orbital
C	l	III	Orientation of Orbital
D	n	IV	orientation of spin of electron

- (1) A-III, B-IV, C-II, D-I    (2) A - II, B - I, C- IV, D - III  
 (3) A-I, B-III, C-II, D-IV    (4) A-III, B-IV, C-I, D-II

**Answer: 4**

67. Match List I with List II. Choose the correct answer from the options given below:

- (1) A-IV, B-I, C-II, D-III      (2) A-I, B-IV, C-II, D-III      (3) A-IV, B-I, C-III, D-II      (4) A-III, B-I, C-II, D-IV

**Answer: 3**

68. Identify the correct reagents that would bring about the following transformation.

- (1) (i)  $BH_3$                       (2) (i)  $H_2O/H^+$                       (3) (i)  $H_2O/H^+$                       (4) (i)  $BH_3$   
 (ii)  $H_2O_2/\overset{\ominus}{O}H$                       (ii)  $PCC$                       (ii)  $CrO_3$                       (ii)  $H_2O_2/\overset{\ominus}{O}H$  (iii)  $PCC$   
 (iii) alk.  $KMnO_4$   
 (iv)  $H_3O^{\oplus}$

**Answer: 4**

69. The reagents with which glucose does not react to give the corresponding tests/products are

- A. Tollen's reagent  
 B. Schiff's reagent  
 C.  $\text{HCN}$   
 D.  $\text{NH}_2\text{OH}$   
 E.  $\text{NaHSO}_3$

Choose the correct options from the given below:

- (1) B and E                      (2) E and D                      (3) B and C                      (4) A and D

**Answer: 1**

70. Choose the correct answer from the options given belows:

	List 1		List 2
A	ethane	I	one $\sigma$ -bond and two $\pi$ -bonds
B	ethene	II	two $\pi$ -bonds
C	carbon molecule	III	one $\sigma$ -bond
D	ethyne	IV	one $\sigma$ -bond and one $\pi$ -bond

- (1) A-III, B-IV, C-II, D-I                      (2) A-III, B-IV, C-I, D-II  
 (3) A-I, B-IV, C-II, D-III                      (4) A-IV, B-III, C-II, D-I

**Answer: 1**

71. Among Group 16 elements, which one does NOT show -2 oxidation state?

- (1)  $\text{Te}$                       (2)  $\text{Po}$                       (3)  $\text{O}$                       (4)  $\text{Se}$

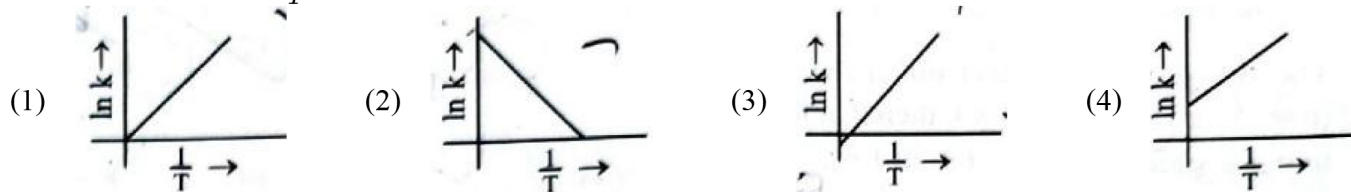
**Answer: 2**

72. For the reaction  $2A \rightleftharpoons B + C$ ,  $K_c = 4 \times 10^{-3}$ . At a given time, the composition of reaction mixture is :  $[A] = [B] = [C] = 2 \times 10^{-3} \text{ M}$ . Then, which of the following is correct?

- (1) Reaction has a tendency to go in backward direction. (2) Reaction has gone to completion in forward direction.  
 (3) Reaction is at equilibrium. (4) Reaction has a tendency to go in forward direction.

**Answer: 1**

73. Which plot of  $\ln k$  vs  $\frac{1}{T}$  is consistent with Arrhenius equation?



**Answer: 2**

74. In which of the following equilibria,  $K_p$  and  $K_c$  are NOT equal?

- (1)  $\text{CO}_{(g)} + \text{H}_2\text{O}_{(g)} \rightleftharpoons \text{CO}_2_{(g)} + \text{H}_2_{(g)}$     (2)  $2\text{BrCl}_{(g)} \rightleftharpoons \text{Br}_2_{(g)} + \text{Cl}_2_{(g)}$     (3)  $\text{PCl}_5_{(g)} \rightleftharpoons \text{PCl}_3_{(g)} + \text{Cl}_2_{(g)}$   
 (4)  $\text{H}_2_{(g)} + \text{I}_2_{(g)} \rightleftharpoons 2\text{HI}_{(g)}$

**Answer: 3**

75. Given below are two statements

Statement I : The boiling point of three isomeric pentanes follows the Order

$n$ -pentane > isopentane > neopentane

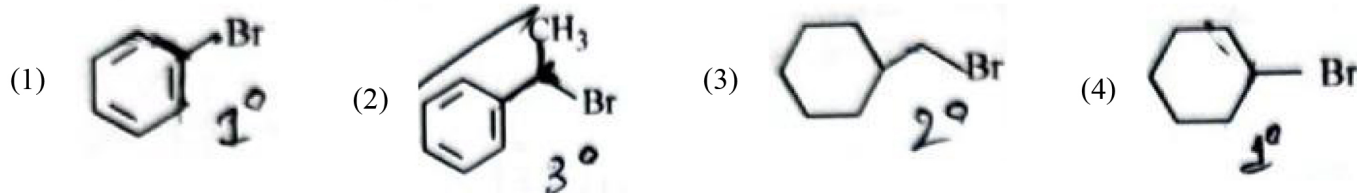
Statement II : When branching increases, the molecule attains a shape of sphere. This results in smaller surface area for contact, due to which the intermolecular forces between the spherical molecules are weak, thereby lowering the boiling point.

In the light of the above statements, choose the most appropriate answer from the options given below:

- (1) Statement I is correct but Statement II is incorrect. (2) Statement I is incorrect but Statement II is correct.  
 (3) Both Statement I and Statement II are correct. (4) Both Statement I and Statement II are incorrect.

**Answer: 3**

76. The compound that will undergo  $S_N$  reaction with the fastest rate is



**Answer: 2**

77. The energy of an electron in the ground state ( $n = 1$ ) for He ion is  $-xJ$ , then that for an electron in  $n = 2$  state for  $Be^{3+}$  ion in  $J$  is :

- (1)  $-4x$  (2)  $-\frac{4}{9}x$  (3)  $-x$  (4)  $-\frac{x}{9}$

**Answer: 3**

78. In which of the following processes entropy increases?

A. A liquid evaporates to vapour.

B. Temperature of a crystalline solid lowered from  $130K$  to  $0K$ .

C.  $2NaHCO_{3(s)} \rightarrow Na_2CO_{3(s)} + CO_{2(g)} + H_2O_{(g)}$

D.  $Cl_{2(g)} \rightarrow 2Cl_{(g)}$  Choose the correct answer from the options given below:

- (1) A, C and D (2) C and D (3) A and C (4) A, B and D

**Answer: 1**

79. On heating, some solid substances change from solid to vapour state without passing through liquid state. The technique used for the purification of such solid substances based on the above principle is known as

- (1) Distillation (2) Chromatography (3) Crystallization (4) Sublimation

**Answer: 4**

80. Match List I with List II.

Choose the correct answer from the options given below:

- (1) A-I, B-IV, C-III, D-II (2) A-II, B-IV, C-III, D-I (3) A-II, B-III, C-IV, D-I (4) A-I, B-III, C-IV, D-II

**Answer: 3**

81. Given below are two statements:

Statement I : Aniline does not undergo Friedel-Crafts alkylation reaction.

Statement II : Aniline cannot be prepared through Gabriel synthesis.

In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I is correct but Statement II is false.      (2) Statement I is incorrect but Statement II is true.  
 (3) Both Statement I and Statement II are true.      (4) Both Statement I and Statement II are false.

**Answer: 3**

82. Arrange the following elements in increasing order of first ionization enthalpy: Li, Be, B, C, N

Choose the correct answer from the options given below:

- (1)  $Li < Be < C < B < N$       (2)  $Li < Be < N < B < C$   
 (3)  $Li < Be < B < C < N$       (4)  $Li < B < Be < C < N$

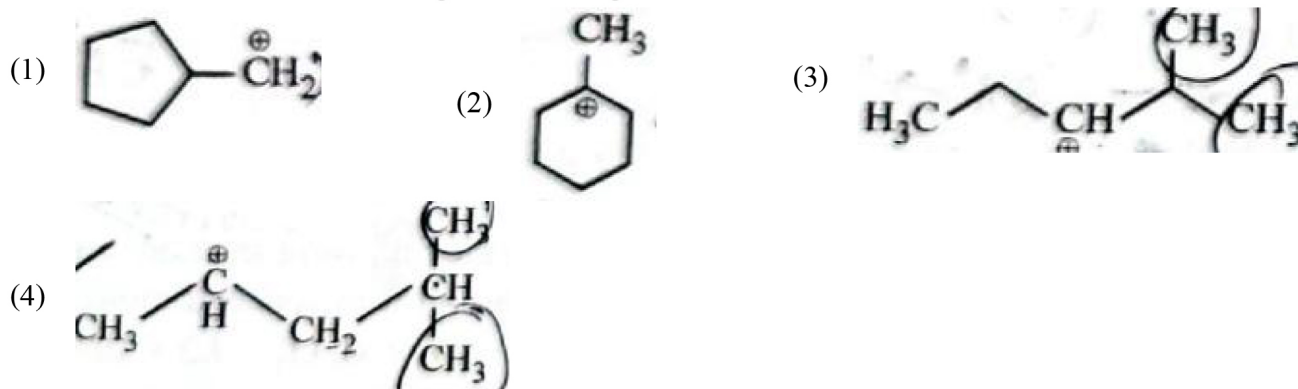
**Answer: 4**

83. The highest number of helium atoms is in

- (1) 4g of helium      (2) 2.271098L of helium at STP      (3) 4mol of helium      (4) 4u of helium

**Answer: 3**

84. The most stable carbocation among the following is:



**Answer: 2**

85. The Henry's law constant ( $K_H$ ) values of three gases (A, B, C) in water are  $145, 2 \times 10^{-5}$  and  $35 \text{ kbar}$ , respectively. The solubility of these gases in water follow the order:

- (1)  $A > C > B$       (2)  $A > B > C$   
 (3)  $B > A > C$       (4)  $B > C > A$

**Answer: 4**

Section - B (Chemistry)

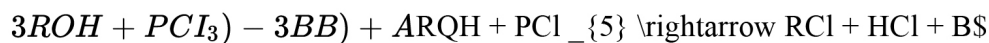
86. A compound X contains 32% of A, 20% of B and remaining percentage of C. Then, the empirical formula of X is :

(Given atomic masses of  $A = 64$ ;  $B = 40$ ;  $C = 32u$ )

- (1)  $AB, C_2$       (2)  $ABC_4$       (3)  $A_2BC_2$       (4)  $ABC_3$

**Answer: 4**

87. The products  $A$  and  $B$  obtained in the following reactions, respectively, are



- (1)  $H_3PO_4$  and  $POCl_3$     (2)  $H_3PO_3$  and  $POCl_3$     (3)  $POCl_3$  and  $H_3PO_3$     (4)  $POCl_3$  and  $H_3PO_4$

**Answer: 2**

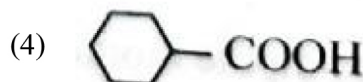
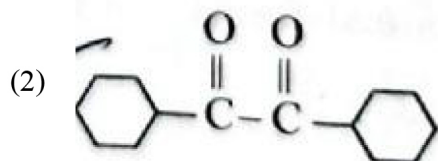
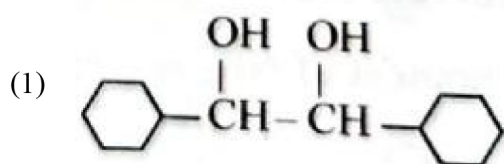
88. The plot of osmotic pressure ( $\Pi$ ) vs concentration ( $\text{mol L}^{-1}$ ) for a solution gives a straight line with slope  $25.73 \text{ L bar mol}^{-1}$ . The temperature at which the osmotic pressure measurement is done is:

(Use  $R = 0.083 \text{ L bar mol}^{-1} \text{ K}^{-1}$ )

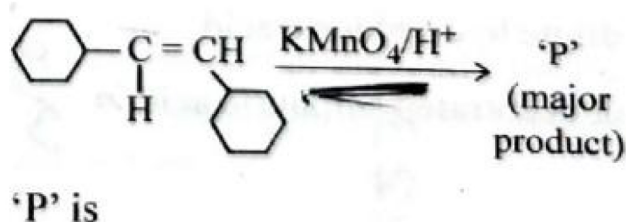
- (1)  $25.73^\circ \text{C}$     (2)  $12.05^\circ \text{C}$     (3)  $37^\circ \text{C}$     (4)  $310^\circ \text{C}$

**Answer: 3**

89. For the given reaction:



**Answer: 4**



90. Given below are two statements : Statement I :  $[Co(NH_3)_6]^{3+}$  is a homoleptic complex whereas  $[Co(NH_3)_4Cl_2]^+$  is a heteroleptic complex.

Statement II : Complex  $[Co(NH_3)_6]^{3+}$  has only

one kind of ligands but  $[Co(NH_3)_4Cl_2]^+$  has more than one kind of ligands: In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I is true but Statement II is false.    (2) Statement I is false but Statement II is true.  
 (3) Both Statement I and Statement II are true.    (4) Both Statement I and Statement II are false.

**Answer: 3**

91. During the preparation of Mohr's salt solution (Ferrous ammonium sulphate), which of the following acids is added to prevent hydrolysis of  $Fe^{2+}$  ion?

- (1) dilute nitric acid    (2) dilute sulphuric acid    (3) dilute hydrochloric acid    (4) concentrated sulphuric acid

**Answer: 2**

92. Identify the correct answer.

- (1) Dipole moment of  $NF_3$  is greater than that of  $NH_3$ .    (2) Three canonical forms can be drawn for  $CO_3^{2-}$  ion.    (3) Three resonance structures can be drawn for ozone.    (4)  $BF_3$  has non-zero dipole moment.

**Answer: 3**

93. Given below are certain cations. qualitative analysis, arrange them in increasing group number from 0 to VI. A.



Choose the correct answer from the options given below:

(1) E, C, D, B, A

(2) E, A, B, C, D

(3) B, A, D, C, E

(4) B, C, A, D, E

**Answer: 3**

94. Identify the major product C formed in the following reaction sequence :

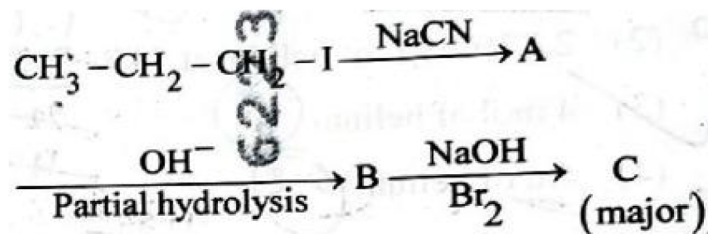
(1) butanamide

(2)  $\alpha$ -bromobutyric acid

(3) propylamine

(4) butylamine

**Answer: 3**



95. The rate of a reaction quadruples when temperature changes from  $27^\circ\text{C}$  to  $57^\circ\text{C}$ . Calculate the energy of activation.

Given  $R = 8.314\text{JK}^{-1}\text{mol}^{-1}$ ,  $\log 4 = 0.6021$

(1)  $3.80\text{kJ/mol}$

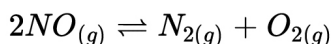
(2)  $3804\text{kJ/mol}$

(3)  $38.04\text{kJ/mol}$

(4)  $380.4\text{kJ/mol}$

**Answer: 3**

96. Consider the following reaction in a sealed vessel at equilibrium with concentrations of  $N_2 = 3.0 \times 10^{-3}\text{M}$ ,  $O_2 = 4.2 \times 10^{-3}\text{M}$  and  $NO = 2.8 \times 10^{-3}\text{M}$



If  $0.1\text{molL}^{-1}$  of  $NO_{(g)}$  is taken in a closed vessel, what will be degree of dissociation ( $\alpha$ ) of  $NO_{(g)}$  at equilibrium?

(1) 0.8889

(2) 0.717

(3)  $0.00889 \times$

(4)  $0.0889 \gg$

**Answer: 1**

97. The work done during reversible isothermal expansion of one mole of hydrogen gas at  $25^\circ\text{C}$  from pressure of 20 atmosphere to 10 atmosphere is:

(Given  $R = 2.0\text{calK}^{-1}\text{mol}^{-1}$ )

(1) 413.14 calories

(2) 100 calories

(3) 0 calorie

(4) -413.14 calorie

**Answer: 4**

98. Mass in grams of copper deposited by passing 9.6487 A current through a voltmeter containing copper sulphate solution for 100 seconds is:

(Given : Molar mass of  $Cu : 63\text{gmol}^{-1}$ ,  $1F = 96487C$ )

(1) 31.5 g

(2) 0.0315 g

(3) 3.15 g

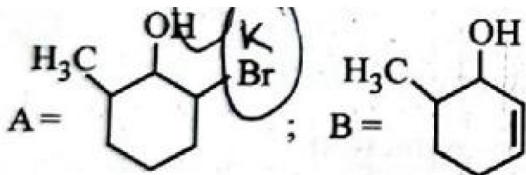
(4) 0.315 g

**Answer: 4**

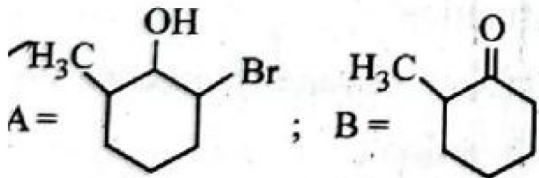


99. Major products A and B formed in the following reaction sequence, are

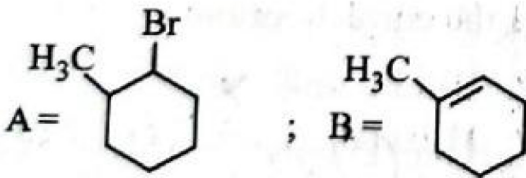
(1)



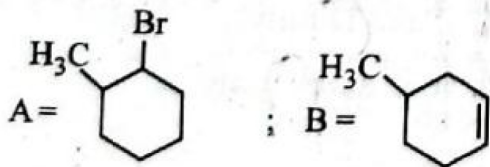
(2)



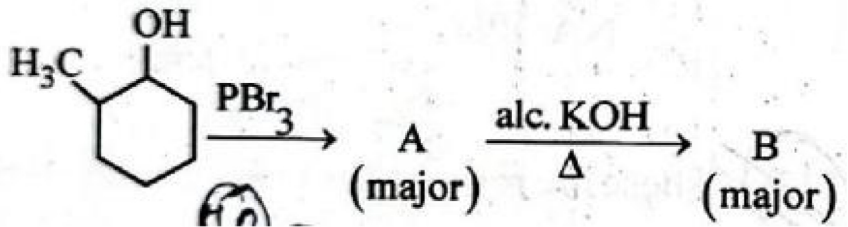
(3)



(4)



Answer: 3



100. The pair of lanthanoid ions which are diamagnetic is

- (1)  $Gd^{3+}$  and  $Eu^{3+}$       (2)  $Pm^{3+}$  and  $Sm^{3+}$       (3)  $Ce^{4+}$  and  $Yb^{2+}$       (4)  $Ce^{3+}$  and  $Eu^{2+}$

Answer: 3

Section - A (Botany)

101. Identify the set of correct statements:

- A. The flowers of Vallisneria are colourful and produce nectar. water  
 B. The flowers of waterlily are not pollinated by water.  
 C. In most of water-pollinated species, the pollen grains are protected from wetting.  
 D. Pollen grains of some hydrophytes are long and ribbon like.  
 E. In some hydrophytes, the pollen grains are carried passively inside water.

Choose the correct answer from the options given below:

- (1) A C , D and E only      (2) B, C, D and E only      (3) C, D and E only      (4) A B, C and D only

Answer: 2

102. The type of conservation in which the threatened species are taken out from their natural habitat and placed in special setting where they can be protected and given special care is called;

- (1) Semi-conservative method (2) Sustainable development (3) in-situ conservation (4) Biodiversity conservation

Answer: 4

103. Inhibition of Succinic dehydrogenase enzyme by malonate is a classical example of:

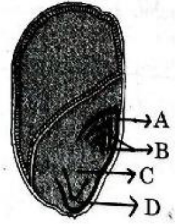
- (1) Competitive inhibition      (2) Enzyme activation      (3) Cofactor inhibition      (4) Feedback inhibition

**Answer: 1**

104. Identify the part of the seed from the given figure which is destined to form root when the seed germinates.

- (1) *C*                                      (2) *D*                                      (3) *A*                                      (4) *B*

**Answer: 1**



105. Bulliform cells are responsible for

- (1) Increased photosynthesis in monocots.                                      (2) Providing large spaces for storage of sugars  
(3) Inward curling of leaves in monocots.                                      (4) Protecting the plant from salt stress.

**Answer: 3**

106. Which of the following is required for the dark reaction of photosynthesis?

A. Light

B. Chlorophyll

C.  $CO_2$

D. ATP

E. NADPH

- Choose the correct answer from the options given below:  
(1) C, D and E only                      (2) D and E only                      (3) A, B and C only                      (4) B, C and D only

**Answer: 1**

107. Formation of interfascicular cambium from fully developed parenchyma cells is an example for

- (1) Dedifferentiation                      (2) Maturation                      (3) Differentiation                      (4) Redifferentiation

**Answer: 1**

108. Hind II always cuts DNA molecules at a particular point called recognition sequence and it consists of:

- (1) 4bp                                      (2) 10bp                                      (3) 8bp                                      (4) 6bp

**Answer: 4**

109. Tropical regions show greatest level of species richness because

- A. Tropical latitudes have remained relatively undisturbed for millions of years, hence more time was available for species diversification.  
B. Tropical environments are more seasonal.  
C. More solar energy is available in tropics.  
D. Constant environments promote niche specialization.  
E. Tropical environments are constant and predictable.

Choose the correct answer from the options given below:

- (1) A, B and E only                      (2) A, B and D only                      (3) A, C, D and E only                      (4) A and B only

**Answer: 3**

110. Which one of the following is not a criterion for classification of fungi?

- (1) Mode of spore formation      (2) Fruiting body      (3) Morphology of mycelium      (4) Mode of nutrition

**Answer: 4**

111. How many molecules of AIF required for every molecule of  $CO_2$  fixedth the Calvin cycle?

- (1) 3 molecules of ATP and 3 molecules of *NADPH*      (2) 3 molecules of ATP and 2 molecules of *NADPH*  
 (3) 2 molecules of ATP and 3 molecufes of *NADPH*      (4) 2 molecules of ATP and 2 molecules of *NADPH*

**Answer: 2**

112. These are regarded as major causes of biodNersity loss:

- A. Over exploitation  
 B. Co-extinction  
 e. Mutation  
 D. Habitat loss and fragmentation  
 E. Migration

Choose the correct option:

- (1) (1) A, B and E only      (2) (2) A, B and Donly      (3) (3) A, C and (Donly      (4) A, B, C and (D) only

**Answer: 2**

113. The capacity to generate a whole plant from any cell of the plant is called:

- (1) Differentiation      (2) Somatic hybridization      (3) Totipotency      (4) Micropropagation

**Answer: 3**

114. The equation of Verhulst-Pearl logistic growth is  $\frac{dN}{dt} = rN \left[ \frac{K-N}{K} \right]$ .

From this equation, *K* indicates:

- (1) Carrying capacity      (2) Population density      (3) Intrinsic rate of natural hiecrease      (4) Biotic potential

**Answer: 1**

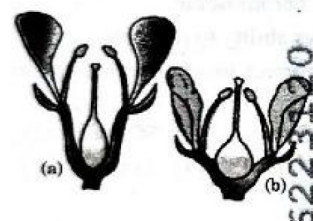
115. Spindle fibers attach to kinetochores of chromosomes during

- (1) Anaphase      (2) Telophase      (3) Prophase      (4) Metaphase

**Answer: 4**

116. Identify the type of flowers based on the position of calyx, corolla and androecifum with respect to the ovary from the given figures (a) and (b)

- (1) (a) Perigynous; (b) Epigynous      (2) (a) Perigynous; (b) Perigynous  
 (3) (a) Epigynous; (b) Hypogynous      (4) (a) Hypogynous; (b) Bppgynous



**Answer: 2**

117. *N* Mist II Mushroom Smut fungus Bread mould Rust fungus Choose the correct answer below:

	List 1		List 2
A	Rhizopus	I	Mushroom
B	Ustilago	II	Smut fungus
C	Puccinia	III	Bread mould
D	Agaricus	IV	Rust fungus

(1) A-III, B-II, C-I, D-IV

(2) A-IV, B-III, C-II, D-I

(3) A-III, B-II, C-IV, D-I

(4) A-I, B-III, C-II, D-IV

**Answer: 3**

118. In a plant, black seed color ( $\overline{BB/Bb}$ ) is dominant over white seed color (bb). In order to find out the genotype of the black seed plant, with which of the following genotype will you cross it?

- (1)  $Bb$  (2)  $BB/Bb$   
 (3)  $BB$  (4)  $bb$

**Answer: 4**

119. A pink flowered Snapdragon plant was crossed with a red flowered Snapdragon plant. What type of phenotype/s is/are expected in the progeny?

- (1) Only pink flowered plants (2) Red, Pink as well as white flowered plants (3) Only red flowered plants  
 (4) Red flowered as well as pink flowered plants

**Answer: 4**

120. Match List I with List II

	List 1		List 2
A	Two or more alternative forms of a gene	I	Back cross
B	Cross of $F_1$ progeny with homozygous recessive parent	II	Ploidy
C	Cross of $F_1$ progeny with any of the parents	III	Allele
D	Number of chromosome sets in plant	IV	test cross

Choose the correct answer from the options given below:

- (1) A-III, B-IV, C-ITD-II (2) A-IV, B-III, C-IN, D-I  
 (3) A-I, B-II, C-III, D-IV (4) A-II, B-I, C-III, D-IV

**Answer: 1**

121. Lecithin, a small molecular weight organic compound found in living tissues, is an example of:

- (1) Glycerides (2) Carbohydrates (3) Amino acids (4) Phospholipids

**Answer: 4**

122. Match List I with List II

	List 1		List 2
A	<i>Clostridium butylicum</i>	I	Ethanol
B	<i>Saccharomyces cerevisiae</i>	II	Streptokinase
C	<i>Trichoderma polysporum</i>	III	Butyric acid
D	<i>Streptococcus</i> sp.	IV	Cyclosporin-A

Choose the correct answer from the options given below:

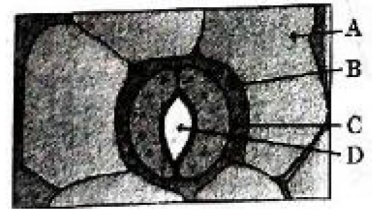
- (1) A-III, B-I, C-IV, D-II (2) A-IV, B-I, C-III, D-II  
 (3) A-III, B-I, C-II, D-IV (4) A-II, B-IV, C-III, D-I

**Answer: 1**

123. In the given figure, which component has thin outer walls and highly thickened inner walls?

- (1) A (2) B (3) C (4) D

**Answer: 3**



124. Which of the following is an example of actinomorphic flower?

- (1) Pisum (2) Sesbania (3) Datura (4) Cassia

**Answer: 3**

125. A transcription unit in DNA is defined prima by the three regions in DNA and these are respect to upstream and down stream end;

- (1) Inducer, Repressor, Structural gene (2) Promotor, Structural.gene, Terminator  
(3) Repressor, Operator gene, Structural ger (4) Structural gene, Transposons, Operatorg

**Answer: 2**

126. What is the fate of a piece of DNA carrying 0 gene of interest which is transferred into an al organism?

- A. The piece of DNA would be able to multi, itself independently  $\hat{\text{A}}\hat{\text{A}}$  the progeny cells the organism.  
B. It may get integrated into the genome of recipient.  
C. It may multiply and be inherited along w the host DNA.  
D. The alien piece of DNA is not an integ part of chromosome.  
E. It shows ability to replicate.

Choose the correct answer from the options giv below:

- (1) B and C only (2) and E only (3) and B only (4) D and E only

**Answer: 1**

127. Auxin is used by gardeners to prepare weed-fto lawns. But no damage is caused to grass as aux

- (1) does not affect mature monocotyledonot plants. (2) can help in cell division in grasses, produce growth.  
(3) promotes apical dominance. (4) promotes abscission of mature leaves onll)

**Answer: 1**

128. The cofactor of the enzyme carboxypeptidase

- (1) Flavin. (2) Haem (3) Zinc (4) Niacin

**Answer: 3**

129. The lactose present in the growth medium bacteria is transported to the cell by the action of

- (1) Permease (2) Polymerase (3) Beta-galactosidase (4) Acetylase

**Answer: 1**

130. Which one of the following can be explained on the basis of Mendel's Law of Dominance?

- A. Out of one pair of factors one is dominant and the other is recessive.
- B. Alleles do not show any expression and both the characters appear as such in  $F_2$  generation.
- C. Factors occur in pairs in normal diploid plants.
- D. The discrete unit controlling a particular character is called factor.
- E. The expression of only (one of the parental characters) is found in a monohybrid cross.

Choose the correct answer from the options given below:

- (1) B, C and D only      (2) A, B, C, D and E      (3) A, B and C only      (4) A, C, D and E only

**Answer: 4**

131. Given below are two statements:

Statement I : Bt toxins are insect group specific and coded by a gene *cry* IAc.

Statement II : Bt toxin exists as inactive protoxin in *B. thuringiensis*. However, after ingestion by the insect the inactive protoxin gets converted into active form due to gastric *pH* of the insect gut. In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I is true but Statement II is false      (2) Statement I is false but Statement II is true  
 (3) Both Statement I and Statement II are true      (4) Both Statement I and Statement II are false

**Answer: 1**

132. Given below are two statements: Statement I : Parenchyma is living but collenchyma is dead tissue.

Statement II : Gymnosperms lack xylem vessels but presence of xylem vessels is the characteristic of angiosperms.

In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I is true but Statement II is false      (2) Statement I is false but Statement II is true  
 (3) Both Statement I and Statement II are true      (4) Both Statement I and Statement II are false

**Answer: 2**

133. Given below are two statements:

Statement I : Chromosome 6 becomes gradually visible under light microscope during leptotene stage.

Statement II : The beginning of diplotene stage is recognized by dissolution of synaptonemal complex. In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I is true but Statement II is false      (2) Statement I is false but Statement II is true  
 (3) Both Statement I and Statement II are true      (4) Both Statement I and Statement II are false

**Answer: 3**

134. Match List I with List II

	List 1		List 2
A	Nucleolus	I	Site of formation of glycolipid
B	Centriole	II	Organization like the cartwheel
C	Leucoplas:	III	Site for active ribosomal RNA synthesis
D	Golgi apparatus	IV	For storing nutrients

Choose the correct answer from the options given below:

- (1) A-III, B-IV, C-II P-1 (2) A-I, B-II, C-III, D-IV  
 (3) A-III, B-II, C-IVCP-I (4) A-II, B-III, C-I, P-IV

**Answer: 4**

135. List of endangered species was released by-

- (1) FOAM (2) IUCN (3) GEAC (4) WWaid

**Answer: 2**

**Section - B (Botany)**

136. The DNA present in chloroplast is:

- (1) Linear, single stranded (2) Circular, single stranded (3) Linear, double stranded (4) Circular, double stranded

**Answer: 4**

137. Which of the following are fused in somatic hybridization involving two varieties of plants?

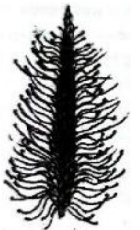
- (1) Protoplasts (2) Pollens (3) Callus (4) Somatic embryos

**Answer: 1**

138. Identify the correct description about the given figure:

- (1) Cleistogamous flowers showing autogamy. (2) Compact inflorescence showing complete autogamy.  
 (3) Wind pollinated plant inflorescence showing flowers with well exposed stamens.  
 (4) Water pollinated flowers showing stamens with mucilaginous covering.

**Answer: 3**



139. Spraying sugarcane crop with which of the following plant growth regulators, increases the length of stem, thus, increasing the yield?

- (1) Cytokinin (2) Absciscic acid (3) Auxin (4) Gibberellin

**Answer: 4**

140. Match List I with List I List II

	<b>List 1</b>		<b>List 2</b>
A	Frederick Griffith	I	Genetic code
B	Francois Jacob & Jacque	II	Semi-conservative mode of DNA replication
C	Har Gobind Khorana	III	Transformation
D	Meselson & Stahl	IV	Lac operon

Choose the correct answer from the options giv below:

(1) A-II, B-III, C-IV, D-I

(2) A-IV, B-I, C-II, D-III

(3) A-III, B-II, C-I, D-IV

(4) A-III, B-IV, C-I, D-II

**Answer: 4**

141. Match List I with List II

	<b>List 1</b>		<b>List 2</b>
A	GLUT-4	I	Hormone
B	Insulin	II	Enzyme
C	Trypsin	III	Intercellular ground substance
D	Collagen	IV	Enables glucose transport into cell:

Choose the correct answer from the options giv below:

(1) A-II, B-III, C-IV, D-I

(2) A-III, B-IV, C-I, D-II

(3) A-IV, B-I, C-II, D-III

(4) {A-I, B-II, C-III,} D-IV

**Answer: 3**

142. Given below are two statements:

Statement I : In  $C_3$  plants, some  $O_2$  binds RuBisCO, hence  $CO_2$  fixation is decreased.

Statement II : In  $C_4$  plants, mesophyll cells show very little photorespiration while bundle sheath cells do not show photorespiration. In the light of the above statements, choose the correct answer from the options given below:

(1) Statement I is true but Statement II is false

(2) Statement I is false but Statement II is true

(3) Both Statement I and Statement II are true

(4) Both Statement I and Statement II are false

**Answer: 3**

143. Identify the step in tricarboxylic acid cycle, which does not involve oxidation of substrate.

(1) Succinyl-CoA  $\rightarrow$  Succinic acid (2) Isocitrate  $\rightarrow$   $\alpha$ -ketoglutaric acid (3) Malic acid  $\rightarrow$  Oxaloacetic acid .(4) Succinic acid  $\rightarrow$  Malic acid**Answer: 1**



## 144. Match List I with List II

	<b>List 1</b>		<b>List 2</b>
A	Citric acid cycle	I	Cytoplasm
B	Glycolysis	II	Mitochondrial matrix
C	Electron transport	III	Intermembrane space of mitochondria
D	Proton gradient	IV	Inner mitochondrial membrane

Choose the correct answer from the option below:

- (1) A-III, B-IV, C-I, D-II                                      (2) A-IV, B-III, C-II, D-I  
 (3) A-I, B-II, C-III, D-IV                                      (4) A-II, B-I, C-IV, D-III

**Answer: 4**

## 145. Which of the following statement is correct regarding the process of replication in E.coli?

- (1) The DNA dependent DNA polymerase catalyses polymerization in  $5' \rightarrow 3'$  as well as  $3' \rightarrow 5'$  direction.  
 (2) The DNA dependent DNA polymerase catalyses polymerization in  $5' \rightarrow 3'$  direction.  
 (3) The DNA dependent DNA polymerase catalyses polymerization in one direction that is  $3' \rightarrow 5'$ .  
 (4) The DNA dependent RNA polymerase catalyses polymerization in one direction, that is  $5' \rightarrow 3'$ .

**Answer: 2**

146. In an ecosystem if the Net Primary Productivity (NPP) of first trophic level is  $100x \text{ (kcalm}^{-2}\text{) } \dot{t}^{-1}$  what would be the GPP (Gross Primary Productivity) of the third trophic level of the same ecosystem?

- (1)  $10x \text{ (kcalm}^{-2}\text{) } yr^{-1}$     (2)  $\frac{100x}{3x} \text{ (kcalm}^{-2}\text{) } yr^{-1}$     (3)  $\frac{x}{10} \text{ (kcalm}^{-2}\text{) } yrr^{-1}$     (4)  $x \text{ (kcalm}^{-2}\text{) } yr^{-1}$

**Answer: 1**

## 147. Match List I with List II

	<b>List 1</b>		<b>List 2</b>
A	Rose	I	Twisted Aestivation
B	Pea	II	Perigynous flower
C	Cotton	III	Drupe
D	Mango	IV	Marginal Placentation

Choose the correct answer from the options given below:

- (1) A-IV, B-III, C-II, D-I                                      (2) A-II, B-III, C-IV, D-I  
 (3) A-II, B-IV, C-I, D-III                                      (4) A-I, B-II, C-III, D-IV

**Answer: 3**

148. Match List I with List II

	List 1		List 2
A	Robert May	I	Species-Area relationship
B	Alexander von Humboldt	II	Long term ecosystem experiment using out door plots
C	Paul Ehrlich	III	Global species diversity at about 7 million
D	David Tilman	IV	Rivet popper hypothesis

Choose the correct answer from the options given below:

- (1) A-I, B-III, C-II, D-IV  
 (2) A-III, B-IV, C-II, D-I  
 (3) A-II, B-III, C-I, D-IV  
 (4) A-III, B-I, C-IV, D-II

**Answer: 4**

149. Match List I with List II Choose the correct anstier from the opti below:

- (1) A-I, B-II, C-IV, B-III    (2) A-III, B-I, C-IV, A-II    (3) A-IV, B-II, C-I,    (4) A-IV, B-I, C-II, P-III

**Answer: 3**

150. Read the following statements and choose the set of correct statements \$In the members of Phagephyceae,

- A. Asexual reproduttion occurs usually by biflagellate zoospres.  
 B. Sexual reproductfon is by oogamous method only.  
 C. Stored food is in the form of carbohydrates which is either mannitol or laminarin.  
 D. The major pigntehts found are chlorophyll a, c and carotenoids and xanthophyll.  
 E. Vegetative cells Shave a cellulosic wall, usually covered on the outside by gelatinous coating of algin.

Choose the correct answer from the options given below:

- (1) A(C)(D) and E onfy    (2) A, B, (C and E onfy    (3) A, B, (C) and C)only \$\\$    (4) B(C) D) and E only

**Answer: 1**

<b>Section - A (Zoology)</b>
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151. Match List I with List II :

	List 1		List 2
A	Typhoid	I	Fungus
B	Leishmaniasis	II	Nematode
C	Ring worm	III	Protozoa
D	Filariasis	IV	Bacteria

- (1) A-III, B-I, C-IV, D-II  
 (2) A-II, B-IV, C-III, D-I  
 (3) A-I, B-III, E-II, D-IV  
 (4) A-IV, B-III, C-I, D-II

**Answer: 4**

152. Match List I with List II :

	List 1		List 2
A	Non - medicated IUD	I	Multiload 375
B	Copper releasing IUD	II	Progestogens
C	Hormone releasing IUD	III	Lippes loop
D	Implants	IV	LNG - 20

Choose the correct answer from the options given below:

- (1) A-IV, B-I, C-II, D-III  
 (2) A-III, B - I, C-IV , D - II  
 (3) A-III, B-I, C-II, D-IV  
 (4) A-I, B-III, C-IV, D-II

**Answer: 2**

153. Given below are the statements :

Statement I : The presence or absence of hymen is not a reliable indicator of virginity.

Statement II : The hymen is torn during the first coitus only.

In the light of the above statements, choose the correct answer from the options given below :

- (1) Statement I is true but Statement II is false  
 (2) Statement I is false but Statement II is true  
 (3) Both Statement I and Statement II are true  
 (4) Both Statement I and Statement II are false

**Answer: 1**

154. In both sexes of cockroach, a pair of jointed filamentous structures called anal cerci are present on :

- (1) 8<sup>th</sup> and 9<sup>th</sup>                      (2) 11<sup>th</sup> segment                      (3) 5<sup>th</sup> segment                      (4) 10<sup>th</sup> segment

**Answer: 4**

155. Match List I with List II :

	List 1		List 2
A	Pons	I	Provides additional space for Neurons, regulates posture and balance.
B	Hypothalamus	II	Controls respiration and gastric secretions.
C	Medulla	III	Connects different regions of the brain.
D	Cerebellum	IV	Neuro secretory cells

Choose the correct answer from the options given below :

- (1) A-I, B-III, C-II, D-IV  
 (2) A-II, B-I, C-III, D-IV  
 (3) A-II, B-III, C-I, D-IV  
 (4) A-III, B-IV, C-II, D-I

**Answer: 4**

156. Which of the following is not a steroid hormone?

- (1) Progesterone                      (2) Glucagon                      (3) Cortisol                      (4) Testosterone

**Answer: 2**

157. Which one is the correct product of DNA dependent RNA polymerase to the given template?

3'TACATGGCAAATATCCATTCA5'

(1) 5'AUGUACCGUUUAUAGGGAAGU3'

(2) 5'ATGTACCGTTTATAGGTAAGT3'

(3) 5'AUGUACCGUUUAUAGGUAAGU3'

(4) 5'AUGUAAAGUUUAUAGGUAAGU3'

**Answer: 3**

158. Three types of muscles are given as *a*, *b* and *c*. Identify the correct matching pair along with their location in human body : skeletal smooth Cardiac Name of muscle/location

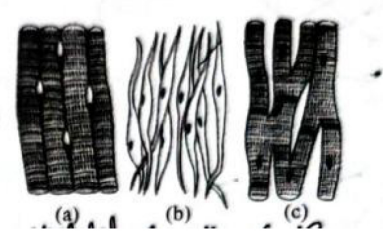
(1) (a) Skeletal - Biceps (b) Involuntary - Intestine (c) Smooth - Heart.

(2) (a) Involuntary - Nose tip (b) Skeletal - Bone (c) Cardiac - Heart.

(3) (a) Smooth - Toes (b) Skeletal - Legs (c) Cardiac - Heart.

(4) (a) Skeletal-Triceps (b) Smooth - Stomach (c) Cardiac - Heart.

**Answer: 4**



159. Following are the stages of cell division :

A. Gap 2 phase

B. Cytokinesis

C. Synthesis phase

D. Karyokinesis

E. Gap 1 phase

Choose the correct sequence of stages from the options given below :

(1) B-D-E-A-C

(2) E-C-A-D-B

(3) C-E-D-A-B

(4) E-B-D-A-C

**Answer: 2**

160. Which of the following are Autoimmune disorders?

A. Myasthenia gravis

B. Rheumatoid arthritis'

C. Gout

D. Muscular dystrophy

E. Systemic Lupus Erythematosus (SLE)

Choose the most appropriate answer from the options given below :

(1) B, C & E only

(2) C, D & E only

(3) A B & D only

(4) A, B & E only

**Answer: 4**

161. Match List I with List II :

	<b>List 1</b>		<b>List 2</b>
A	Lipase	I	Peptide bond
B	Nuclease	II	Ester bond
C	Protease	III	Glycosidic bond
D	Cerebellum	IV	Phosphodiester bond

Choose the correct answer from the options given below

(1) A-II, B-IV, CAI, D-III

(2) A-IV, B-I, C-III, D-I

(3) A-IV, B-II, EVIII, D-I.

(4) A-III, B-II, C-I, D - IV

**Answer: 1**

162. The flippers of the Penguins and Dolphins are the example of the

(1) Convergent evolution

(2) Divergent evolution

(3) Adaptive radiation

(4) Natural selection

**Answer: 1**

163. Match List I with List II :

	<b>List 1</b>		<b>List 2</b>
A	Expirator capacity	I	Expiratory reserve volume + Tidal volume + Inspiratory reserve volume
B	Functional residual capacity	II	Tidal volume + Expiratory reserve volume
C	Vital capacity	III	Tidal volume + Inspiratory reserve volume
D	Inspiratory capacity	IV	Expiratory reserve volume + Residual volume

Choose the correptanswer from the options given below :

(1) A-II, B-I, C-IN, D-III

(2) A-I, B-III, QGI, D-IV

(3) A-II, B-IV, C-I, D-III

(4) A-III, B-II,C-IV, D-I

**Answer: 3**

164. Which one of the following factors will not affect the Hardy-Weinberg equilibrium?

(1) Gene migration

(2) Constant gene pool

(3) Genetic recombination

(4) Genetic drift

**Answer: 2**

165. Given below are some stages of human evolution Arrange them in correct sequence. ( Past to Recent)

- A. Homo habiliss
- B. Homo sapiens
- C. Homo neanderthalensis
- D. Home erectus

Choose the correct sequence of human evolution from the options given below :

- (1) C-B-D-A                      (2) A-D-C-B                      (3) D-A-C-B                      (4) B-A-D-C

**Answer: 2**

166. Following are the stages of pathway for conduction of an action potential through the heart:

- A. AV bundle
- B. Purkinje fibres
- C. AV node
- D. Bundle branches
- E. SA node

Choose the correct sequence of pathway from the options given below :

- (1) B-D-E-C-A                      (2) E-A-D-B-C                      (3) E-C-A-D-B                      (4) A-E-C-B-D

**Answer: 3**

167. Which of the following factors are favourable for the formation of oxyhaemoglobin in alveoli?

- (1) Low  $pCO_2$  and High  $H^+$  concentration    (2) Low  $pCO_2$  and High temperature    (3) High  $pO_2$  and High  $pCO_2$   
 (4) High  $pO_2$  and Lesser  $H^+$  concentration

**Answer: 4**

168. Match List I with List II :

	<b>List 1</b>		<b>List 2</b>
A	$\alpha - 1$ antitrypsin	I	Cotton bollworm
B	Cry IAb	II	ADA deficiency
C	Cry IAc	III	Emphysema
D	Enzyme replacement therapy	IV	Corn borer

Choose the correct answer from the options given below :

- (1) A-III, B-IV, C-I, D-II                      (2) A-II, B-IV, C-IV D-III  
 (3) A-II, B-I, C-IV, D-III                      (4) A-II, B-I, C-II, D-IV

**Answer: 1**

169. Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R :

Assertion A : FSH acts upon ovarian follicles in female and Leydig cells in male.

Reason R : Growing ovarian follicles secrete estrogen in female while interstitial cells secrete androgen in male human being.

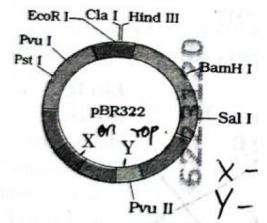
In the light of the above statements, choose the correct answer from the options given below :

- (1) A is true but R is false (2) A is false but R is true  
 (3) Both A and R are true and R is the correct explanation of A.  
 (4) Both A and R are true but R is NOT the correct explanation of A.

**Answer: 2**

170. The following diagram showing restriction sites in E.coli cloning vector pBR322. Find the role of ' X ' and ' Y ' genes :

- (1) The gene ' X ' is for protein involved in replication of Plasmid and ' Y ' for resistance to antibiotics.  
 (2) Gene ' X ' is responsible for recognition sites and ' Y ' is responsible for antibiotic resistance.  
 (3) The gene ' X ' is responsible for resistance to antibiotics and ' Y ' for protein involved in the replication of Plasmid.  
 (4) The gene ' X ' is responsible for controlling the copy number of the linked DNA and ' Y ' for protein involved in the replication of Plasmid.



**Answer: 4**

171. Match List I with List II : List I Choose the correct answer from the op below :

- (1) A-II, B-I, C-III, D-IV (2) A-III, B-IV, C-I, D-II (3) A-IV, B-III, C-I, D-II (4) A-I, B-III, C-II, D-IV

**Answer: 2**

172. Consider the following statements :

A. Annelids are true coelomates

B. Poriferans are pseudocoelomates >

C. Aschelminthes are acoelomates D. Platyhelminthes are pseudocoelomates Choose the correct answer from the options given below :

- (1) C only (2) D only (3) B only (4) A only

**Answer: 4**

173. Given below are two statements : Statement I : In the nephron, the descending limb of loop of Henle is impermeable to water and permeable to electrolytes. Statement II : The proximal convoluted tubule is lined by simple columnar brush border epithelium and increases the surface area for reabsorption. In the light of the above statements, choose the correct answer from the options given below :

- (1) Statement I is true but Statement II is false (2) Statement I is false but Statement II is true  
 (3) Both Statement I and Statement II are true (4) Both Statement I and Statement II are false

**Answer: 4**

174. Match List I with List II :

	<b>List 1</b>		<b>List 2</b>
A	Fibrous joints	I	Adjacent vertebrae, limited movement
B	Cartilaginous joints	II	Humerus and Pectoral girdle, rotational movement
C	Hinge joints	III	Skull, don't allow any movement
D	ball and socket joints	IV	Knee, help in locomotion

Choose the correct answer from the options given below :

(1) A-II, B-III, C-I, D-IV

(2) A-III, B-I, C-IV, D-II

(3) A-IV, B-II, C-III, D-I

(4) A-I, B-III, C-II, D-IV

**Answer: 2**

175. Which of the following is not a natural/traditional contraceptive method?

(1) Lactational amenorrhea

(2) Vaults

(3) Coitus interruptus

(4) Periodic abstinence

**Answer: 2**

176. Match List I with List II :

	<b>List 1</b>		<b>List 2</b>
A	Pleurobraciths	I	Mollusca
B	Radul	II	Ctenophora
C	Stomochord	III	Osteichthyes
D	Air bladder	IV	Hemichordata

Choose the correct answer from the options given below :

(1) A-II, B-IV, C-I, D-III

(2) A-IV, B-III, C-II, D-I

(3) A-IV, B-II, C-III, D-I

(4) A-II, B-I, C-IV, D-III

**Answer: 4**

177. Match List I with List II :

	<b>List 1</b>		<b>List 2</b>
A	Axoneme	I	Centriole
B	Cartwheel pattern	II	Cilia and flagella
C	Cristar	III	Chromosome
D	Satellite	IV	Mitochondria

Choose the correct answer from the options given below :

(1) A-II, B-IV, C-I, D-III

(2) A - II , B - I , C - IV ,D - III

(3) A-IV, B-III, C-II, D-I

(4) A-IV, B-II, C-III, D-I

**Answer: 2**



178. Which of the following statements is incorrect?

- (1) Bio-reactors are used to produce smallscale bacterial cultures.
- (2) Bio-reactors have an agitator system, an oxygen delivery system and foam control system.
- (3) A bio-reactor provides optimal growth conditions for achieving the desired product.
- (4) Most commonly used bio-reactors are of stirring type.

**Answer: 1**

179. Match List I with List II Choose the correct answer from the options given below:

- (1) A-II, B-IV, C-I, D-III    (2) A-IV, B-III, C-II, D-I    (3) A-IV, B-II, C-III, D-I    (4) A-I, B-II, C-IV, D-III

**Answer: 1**

180. Match List I with List II :

	List 1		List 2
A	Diakinesis	I	Synaptonemal complex formation
B	Pachytene	II	Completion of terminalisation of chiasmata
C	Zygotene	III	Chromosomes look like thin threads
D	Leptotene	IV	Appearance of recombination nodules

Choose the correct answer from the options given below;

- (1) A-III, B-I, C-II, D-IV                      (2) A-IV, B-II, C-III, D-I  
(3) A-II, B-IV, C-III, D-I                      (4) A-I, B-III, C-II, D=IV

**Answer: 1**

181. Given below are two statements : one is labelled as Assertion  $A$  and the other is labelled as Reason  $R$  :

Assertion  $A$ : Breast-feeding during initial period of infant growth is recommended by doctors for bringing a healthy baby.

Reason  $R$  : Colostrum contains several antibodies absolutely essential to develop resistance for the new born baby.

In the light of the above statements, choose the most appropriate answer from the options given below :

- (1)  $A$  is correct but  $R$  is not correct.                      (2)  $A$  is not correct but  $R$  is correct.  
(3) Both  $A$  and  $R$  are correct and  $R$  is the correct explanation of  $A$ .  
(4) Both  $A$  and  $R$  are-correct but  $R$  is NOT the correct explanation of  $A$ .

**Answer: 3**

182. Match List I with List II:

Choose the correct answer from the options given below :

- (1) A-IV, B-I, C-II, D-III    (2) A-III, B-II, C-I, D-IV    (3) A-II, B-I, C-III, D-IV    (4) A-III, B-I, C-II, D-IV

**Answer: 4**

183. The "Ti plasmid" of *Agrobacterium tumefaciens* stands for

- (1) Tumor inducing plasmid    (2) Temperature independent plasmid    (3) Tumour;inhibiting plasmid  
(4) Tumor independent plasmid

**Answer: 1**

184. Which of the following is not a component of Fallopian tube?

- (1) Infundibulum (2) Ampulla (3) Uterine fundus (4) Isthmus

**Answer: 3**

185. Match List I with

Choose the correct answer from the options given below :

- (1) A-III, B-IV, C-I, D-II (2) A-IV, B-I, C-II, D-III (3) A-I, B-II, C-III, D-IV (4) A-II, B-III, C-IV, D-I

**Answer: 1**

**Section - B (Zoology)**

186. The following are the statements about nonchordates:

- A. Pharynx is perforated by gill slits.  
B. Notochord is absent.  
C. Central nervous system is dorsal.  
D. Heart is dorsal if present.  
E. Post anal tail is absent.

Choose the most appropriate answer from the options given below :

- (1) B,D&E Only (2) B, C & D only (3) A & C only (4) A,B& D only

**Answer: 1**

187. Match List I with List II : Choose the correct answer from the options given below :

- (1) A-I, B-II, C-IV, D-III (2) A-III, B-I, C-IV, D-II (3) A-II, B-I, C-III, D-IV (4) A-III, B-I, C-II, D-IV

**Answer: 2**

188. Given below are two statements : Statement I : The cerebral hemispheres are connected by nerve tract known as corpus callosum. Statement II : The brain stem consists of the medulla oblongata, pons and cerebrum. In the light of the above statements, choose the most appropriate answer from the options given below :

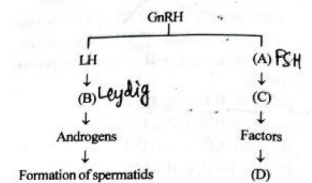
- (1) Statement I is correct but Statement II is incorrect. (2) Statement I is incorrect but Statement II is correct.  
(3) Both Statement I and Statement II are correct. (4) Both Statement I and Statement II are incorrect.

**Answer: 1**

189. Identify the correct option (A), (B), (C), (D) with respect to spermatogenesis.

- (1) (1) FSH, Sertoli cells, Leydig cells, spermatogenesis.  
(2) (2) ICSH, Leydig cells, Sertoli cells, spermatogenesis.  
(3) (3) FSH, Leydig cells, Sertoli cells, spermiogenesis  
(4) (4) ICSH, Interstitial cells, Leydig cells, spermiogenesis.

**Answer: 3**



190. Match List I with List II : Choose the correct answer from the options given below :

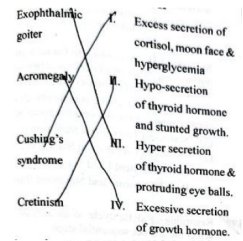
- (1) A-III, B-IV, C-I, D-II (2) A-IV, B-III, C-I, D-II (3) A-II, B-IV, C-I, D-III (4) A-III, B-II, C-IV, D-I

**Answer: 2**

191. Match List I with List II : Choose the correct answer from the options given below :

- (1) A-III, B-IV, C-II, D-I      (2) A-III, B-IV, C-I, D-II      (3) A-I, B-III, C-II, D-IV  
 (4) A-IV, B-II, C-I, D-III

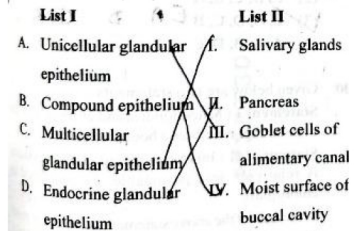
**Answer: 2**



192. Match List I with List II : Choose the correct answer from the options given below :

- (1) A-III, B-IV, C-I, D-II      (2) A-II, B-I, C-IV, D-III      (3) A-II, B-I, C-III, D-IV  
 (4) A-IV, B-III, C-I, D-II

**Answer: 1**



193. Given below are two statements : Statement I : Bone marrow is the main lymphoid organ where all blood cells including lymphocytes are produced. Statement II : Both bone marrow and thymus provide micro environments for the development and maturation of T-lymphocytes. In the light of the above statements, choose the most appropriate answer from the options given below:

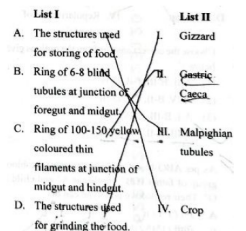
- (1) Statement I is correct but Statement II is incorrect.      (2) Statement I is incorrect but Statement II is correct.  
 (3) Both Statement I and Statement II are correct.      (4) Both Statement I and Statement II are incorrect.

**Answer: 3**

194. Match List I with List II related to digestive system of cockroach. Choose the correct answer from the options given below:

- (1) A-IV, B-III, C-II, D-I      (2) A-III, B-II, C-IV, D-I      (3) A-IV, B-II, C-III, D-I  
 (4) A-I, B-II, C-III, D-IV

**Answer: 3**



195. Choose the correct statement given below regarding juxta medullary nephron.

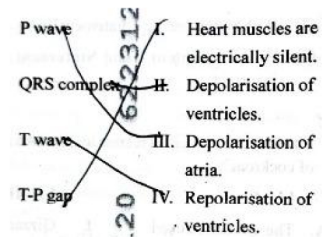
- (1) Loop of Henle of juxta medullary nephron runs deep into medulla.  
 (2) Juxta medullary nephrons outnumber the cortical nephrons.  
 (3) Juxta medullary nephrons are located in the columns of Bertini.  
 (4) Renal corpuscle of juxta medullary nephron lies in the outer portion of the renal medulla.

**Answer: 1**

196. Match List I with List II : Choose the correct answer from the options given below :

- (1) A-II, B-III, C-I, D-IV      (2) A-IV, B-II, C-I, D-III      (3) A-I, B-III, C-IV, D-II  
 (4) A-III, B-II, C-IV, D-I

**Answer: 4**



197. As per ABO blood grouping system, the blood group of father  $B^+$ , mother is  $A^+$  and child is  $O^+$ . Their respective genotype can be

A.  $I^{B_i}/I^{A_i}/i_i$

B.  $I^B I^B / I^A I^A / ii$

C.  $I^A I^B / i I^A / I^{B_i}$

D.  $I^{A_i} / I^{B_i} / I^A$

E.  $i I^B / I I^A / I I^B$  Choose the most appropriate answer from the options given below :

(1) C & B only

(2) D & E only

(3) A only

(4) B only

**Answer: 1**

198. Given below are two statements :

Statement I : Gause's competitive exclusion principle states that two closely related species, competing for different resources cannot exist indefinitely? Statement II : According to Gause's principle, during competition, the inferior will be eliminated. This may be true if resources are limiting. In the light of the above statements, choose correct answer from the options given below:

(1) Statement I is true but Statement II is false

(2) Statement I is false but Statement II is true

(3) Both Statement I and Statement II are true

(4) Both Statement I and Statement II are false

**Answer: 2**

199. Regarding catalytic cycle of an enzyme action select the correct sequential steps :

A. Substrate enzyme complex formation.

B. Free enzyme ready to bind with another substrate.

C. Release of products.

D. Chemical bonds of the substrate broken.

E. Substrate binding to active site.

Choose the correct answer from the options given below :

(1) B, A, C, D, E

(2) E, D, C, B, A

(3) E, A, D, C, B

(4) A, E, B, D, C,

**Answer: 3**

200. Given below are two statements :

Statement I: Mitochondria and chloroplasts are both double membrane bound organelles.

Statement II : Inner membrane of mitochondria is relatively less permeable, as compared to chloroplast.

In the light of the above statements, choose the most appropriate answer from the options given below

(1) Statement I is correct but Statement II is incorrect.

(2) Statement I is incorrect but Statement II is correct

(3) Both Statement I and Statement II are correct

(4) Both Statement I and Statement II are incorrect.

**Answer: 1**